APPENDIX BTECHNICAL DRAWINGS AND

CHEMICAL INVENTORY

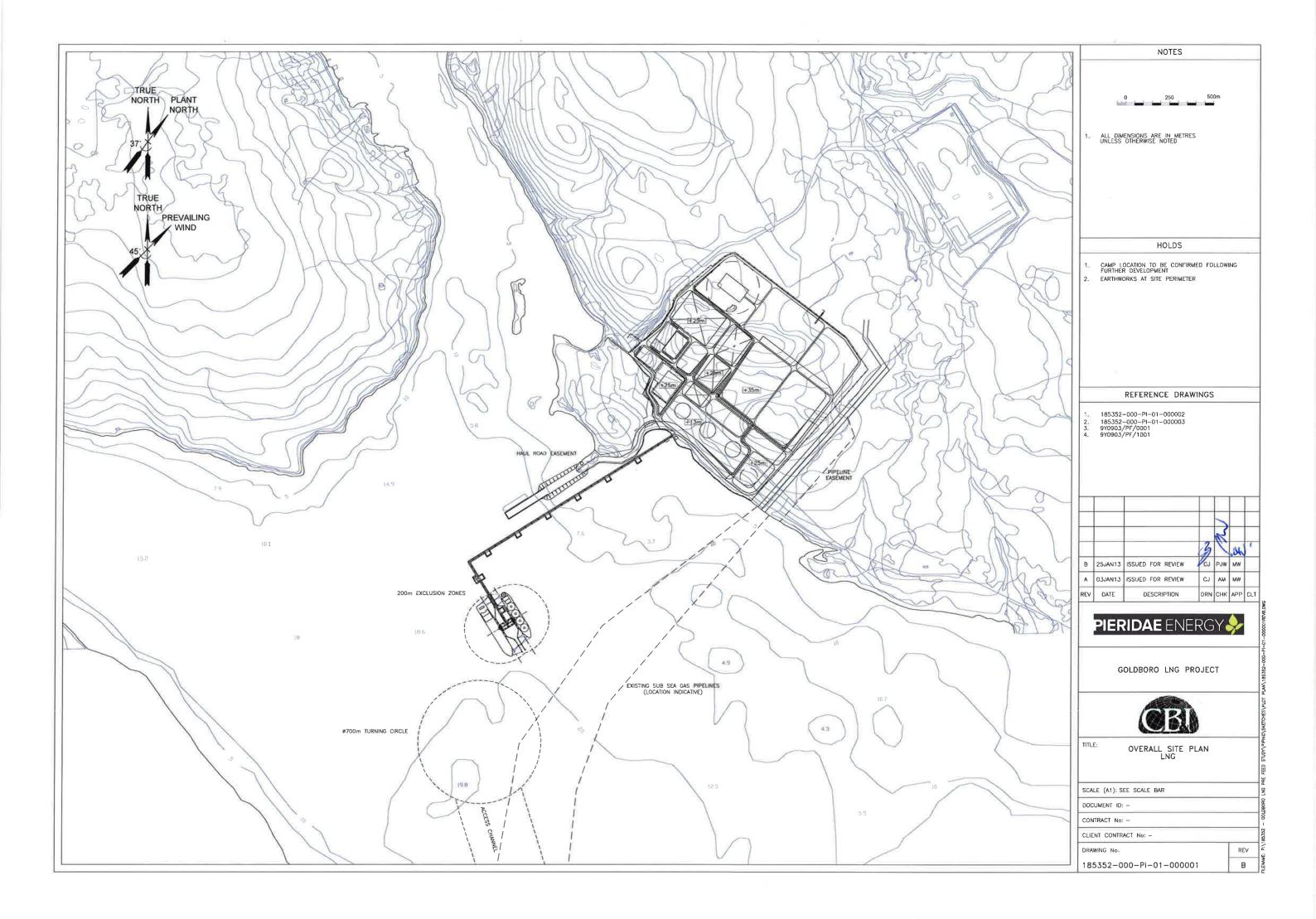




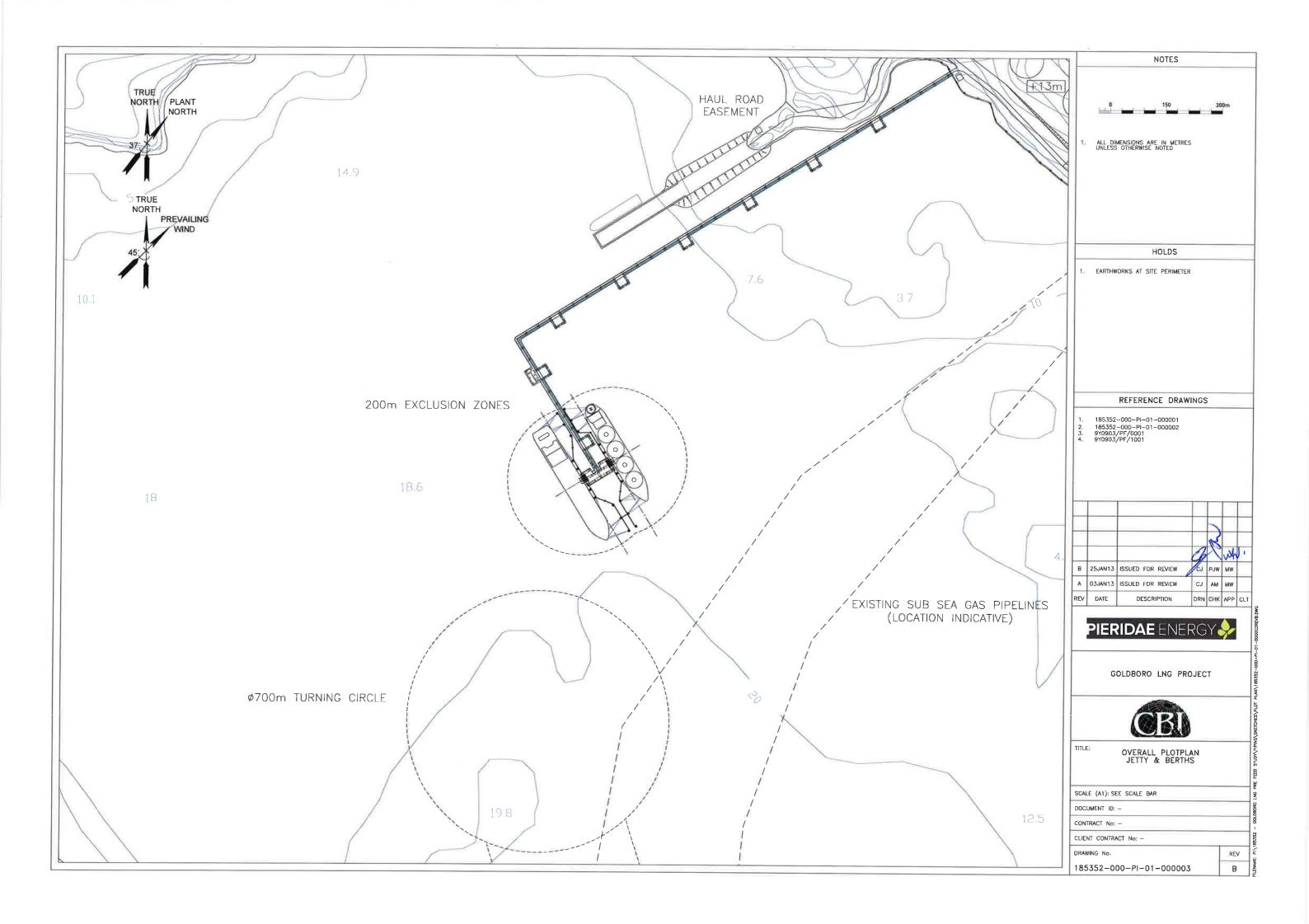
Appendix B-1: Goldboro LNG Facility Technical Drawings

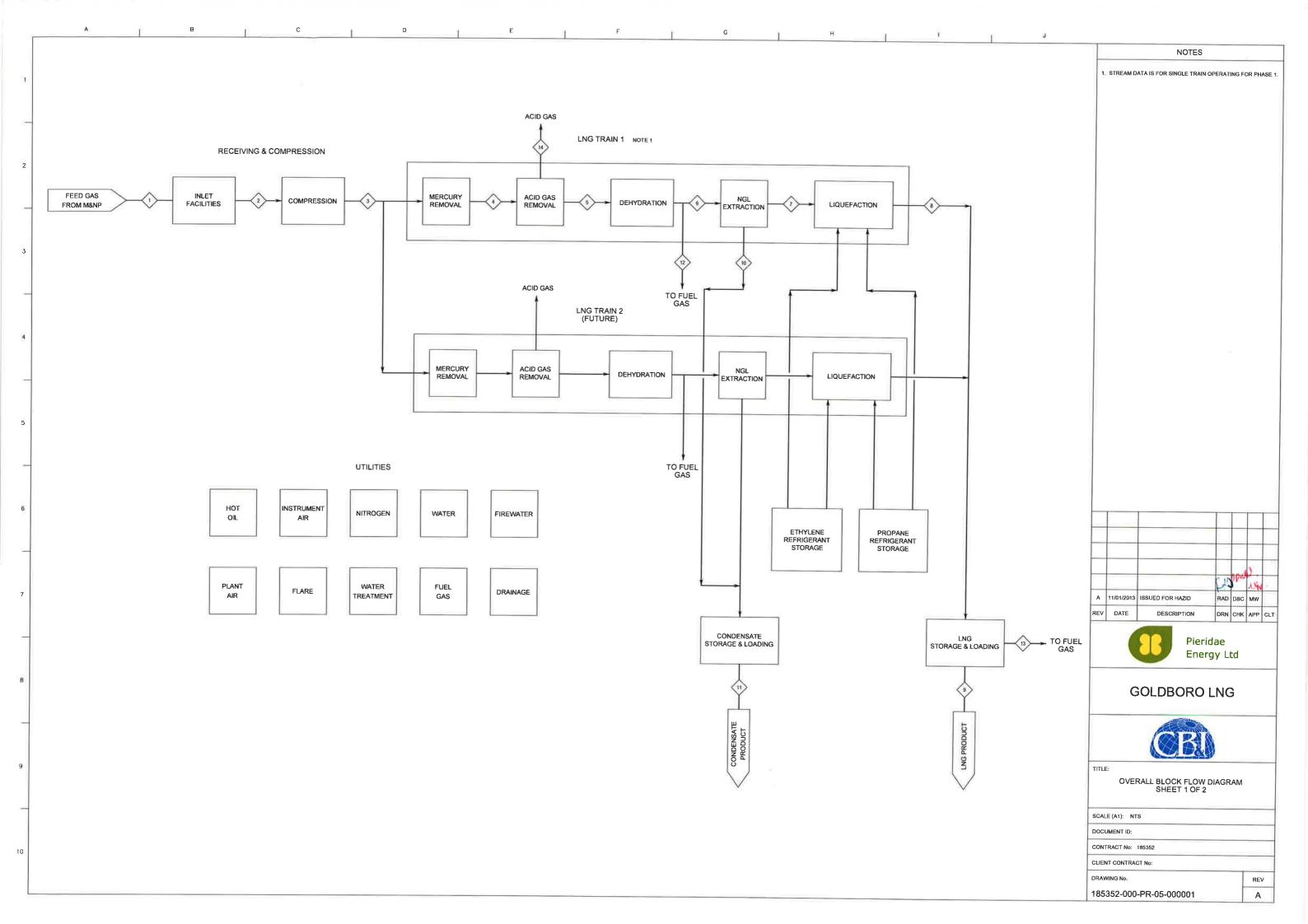
List of Drawings:

Overall Site Plan - LNG
Overall Plot Plan - LNG
Overall Plot Plan - Jetty & Berths
LNG Jetty - Approach Trestle Details
Overall Block Flow Diagram (2 sheets)
Generic Liquefaction Schematic









Stream No.		1	2	3	4	5	6	7	8	9	10	11	12	13	14
Temperature	°C	7.0	7.0	45.0	44.7	44.9	44.5	73.9	-151.4	-160,5	44.5	44.5	44.5	60.0	1100.0
Pressure	barg	43.0	43,0	66.0	65.2	64.6	63.0	67.7	10.0	0.1	63,0	63.0	63.0	30.0	0.0
Phase		Vapour	Liquid	Liquid	Liquid	Liquid	Vapour	Vapour	Vapour						
Molecular weight		17.6	17.6	17.6	17.6	17,2	17.2	17.2	17.2	17.2	78.2	78.2	17.2	16.5	44.0
Density	kg/m³	38.3	38.3	50.6	50.0	48.2	47.0	44.7	432.7	446.2	627,2	627.2	47.0	19.3	0.4
Mass Flow	kg/h	702414	702414	702414	702414	673170	655465	655246	655246	604580	219	219	17705	50666	32326
Mass Flow (Note 3)	MTpa	5.7	5.7	5.7	5.7	5.5	5.3	5.3	5.3	4.9	0.0	0.0	0.1	0.4	0.3
Notes										4, 5		5		5	2

NOTES

- 1, STREAM DATA IS FOR SINGLE TRAIN OPERATING FOR PHASE 1, 2, ACID GAS VENT INCLUDES FOR COMBUSTION OF FUEL GAS IN
- ACID GAS INCINERATOR. 3, ANNUAL PRODUCTION RATES ARE BASED ON AN ASSUMED

FACILITY AVAILABILITY OF 340 DAYS/YEAR

- 4, LNG PRODUCT STREAM IS A ESTIMATED FREE ON BOARD VALUE ALLOWING FOR EXPECTED STORAGE AND LOADING LOSSES.
- 5, AGGREGATE RATE BASED ON A TIME WEIGHTED AVERAGE OF ESTIMATED HOLDING AND LOADING MODE OPERATION.

HOLDS

1. HOLD FOR LIQUEFACTION LICENSOR INPUT.

11/01/2013 ISSUED FOR REVIEW DESCRIPTION DATE



Energy Ltd

GOLDBORO LNG



TITLE:

OVERALL BLOCK FLOW DIAGRAM SHEET 2 OF 2

SCALE (A1): NTS

DOCUMENT ID:

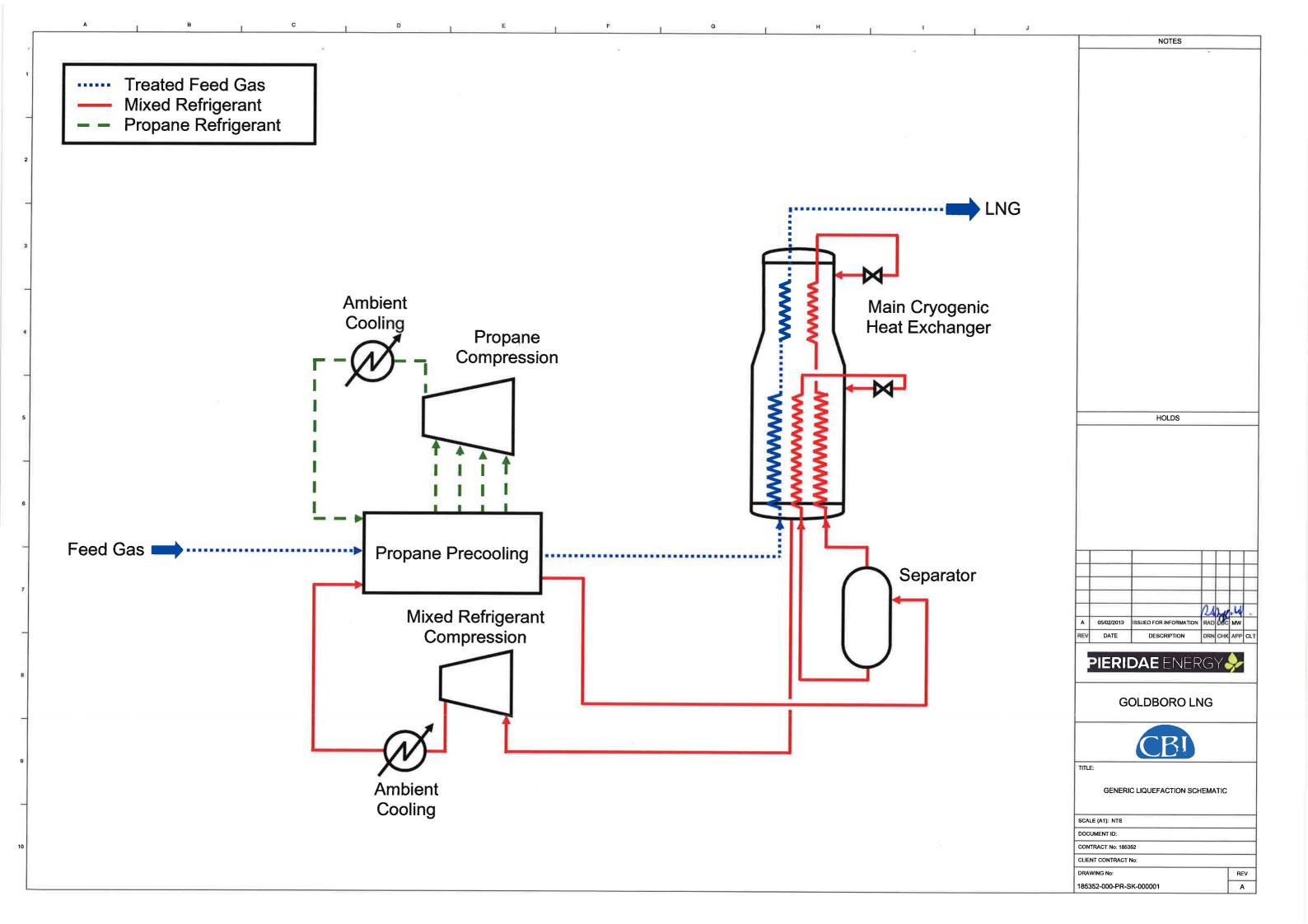
CONTRACT No: 185352

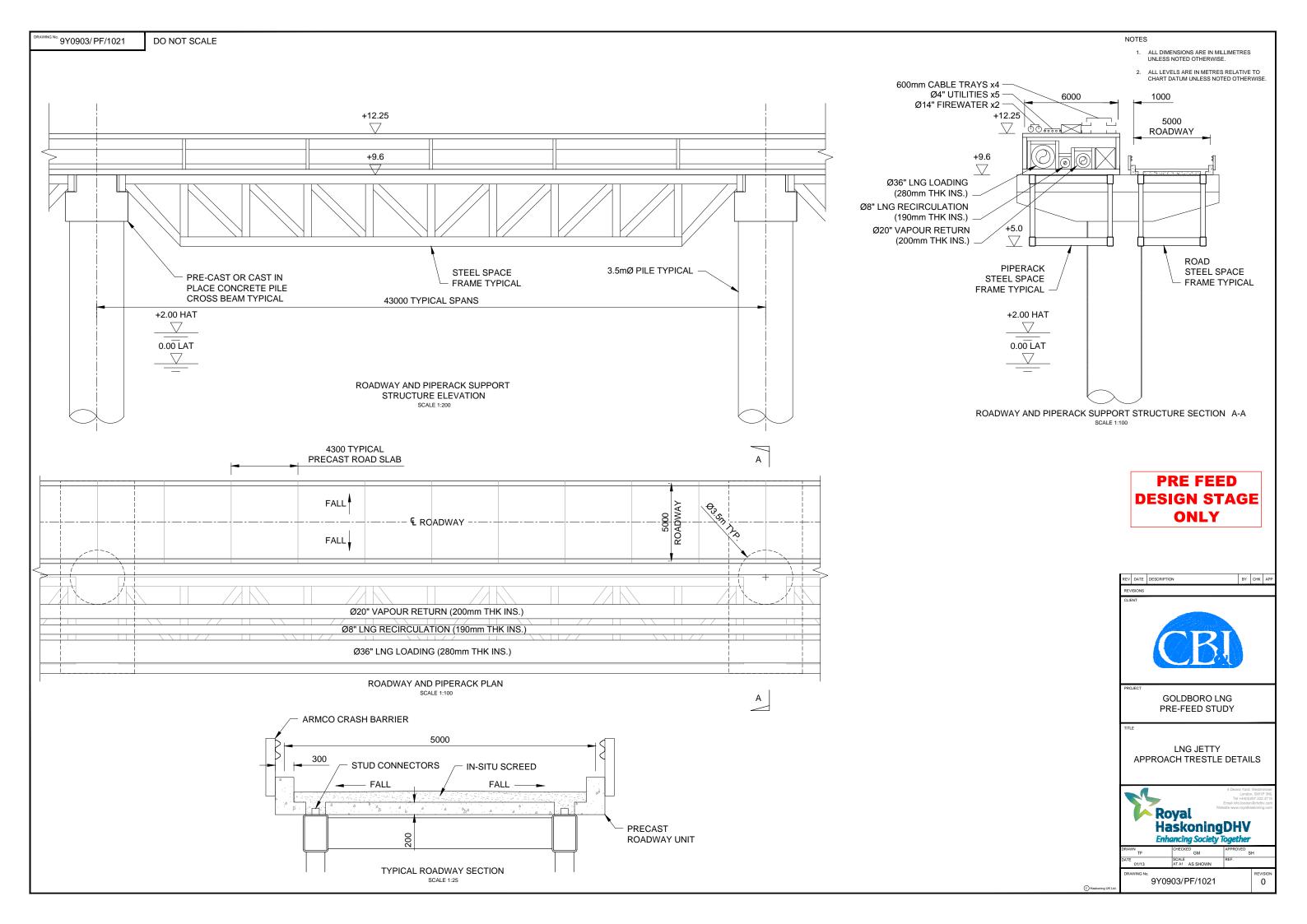
CLIENT CONTRACT No:

DRAWING No: 185352-000-PR-05-000001

Α

REV







Appendix B-2: Water Intake Figures

List of Figures:

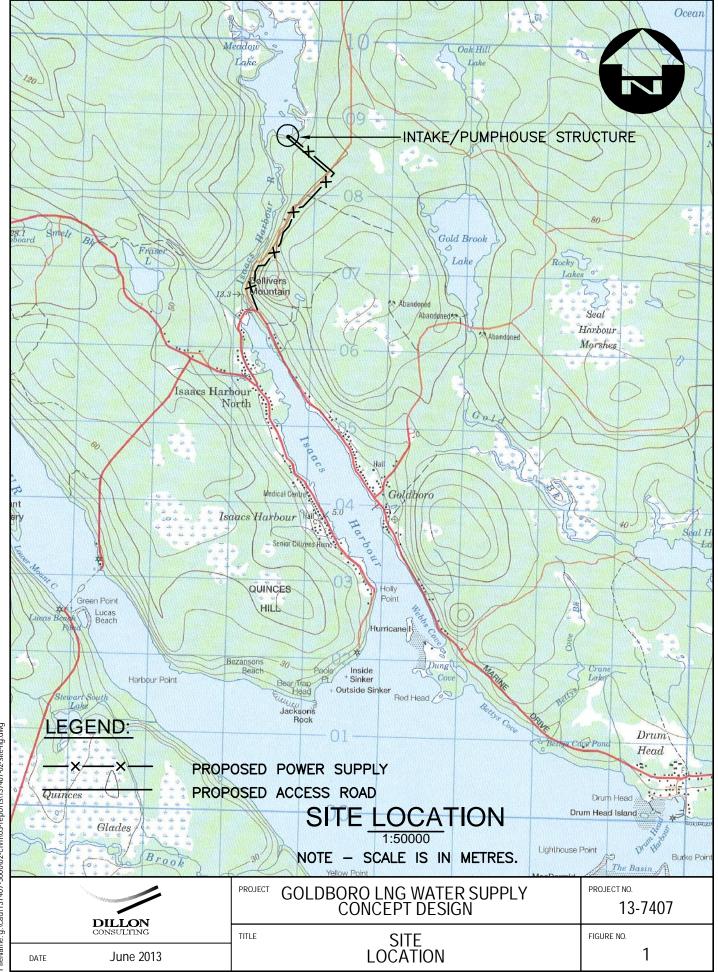
Figure 1 Site Location

Figure 2 Water Intake Structure Profile

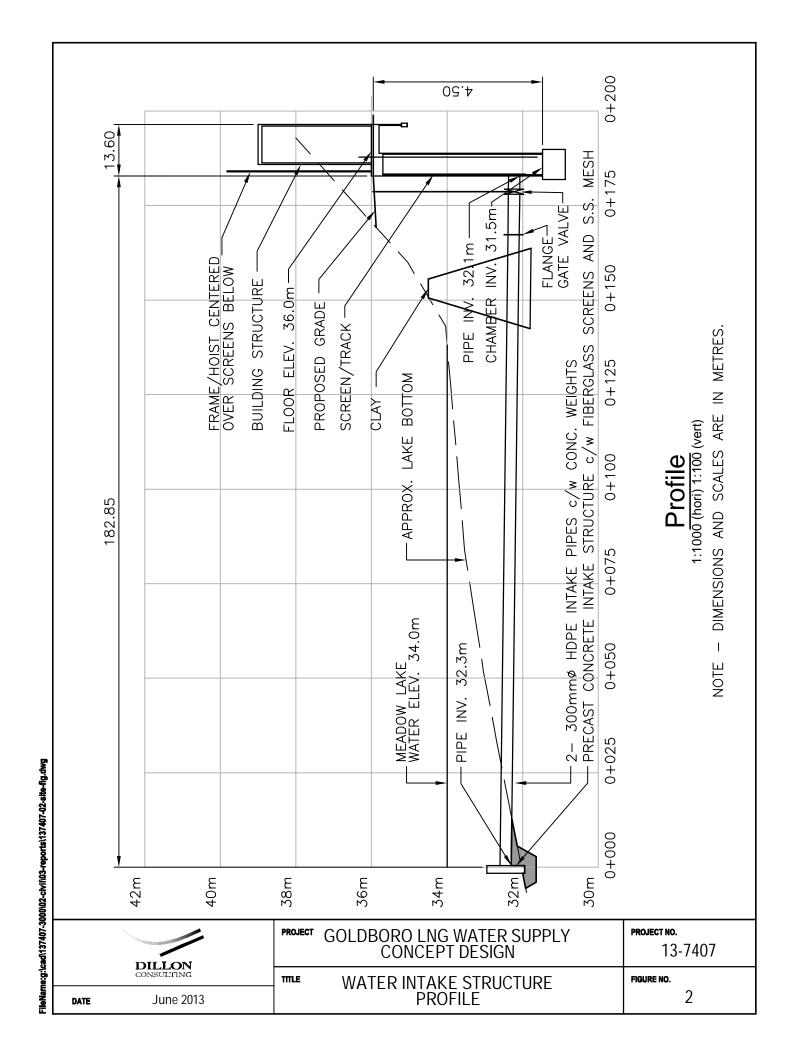
Figure 3 Water Intake Structure Plan

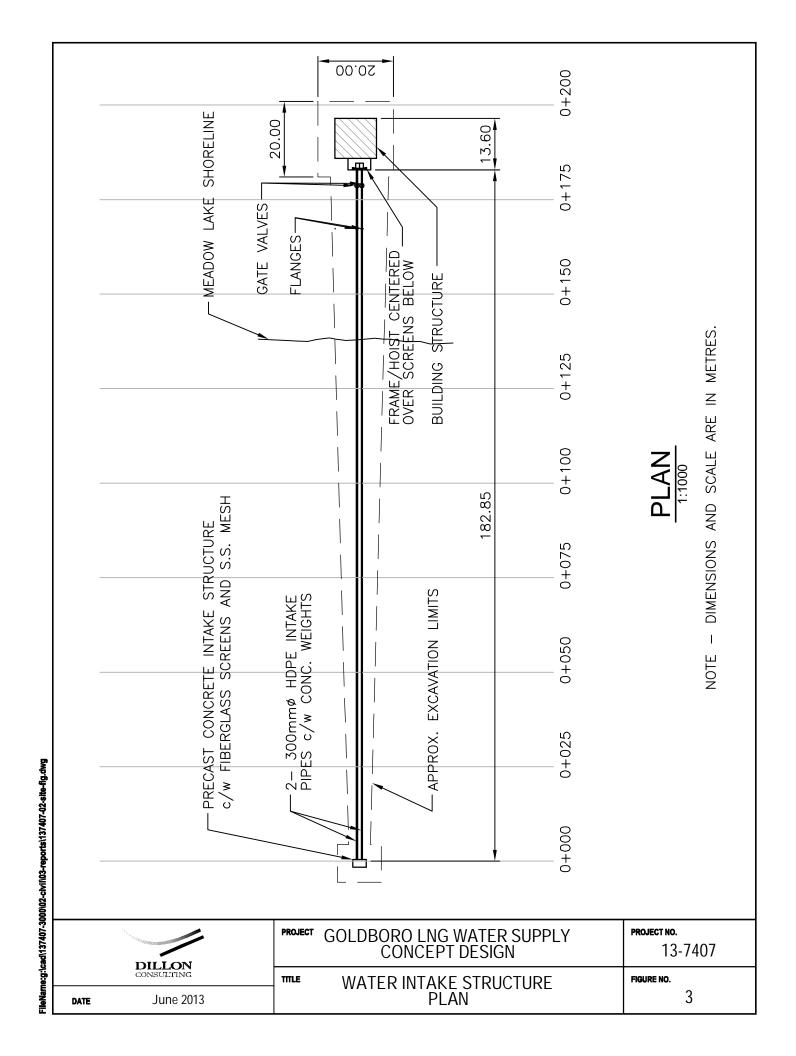
Figure 4 Pump House Floor Plan

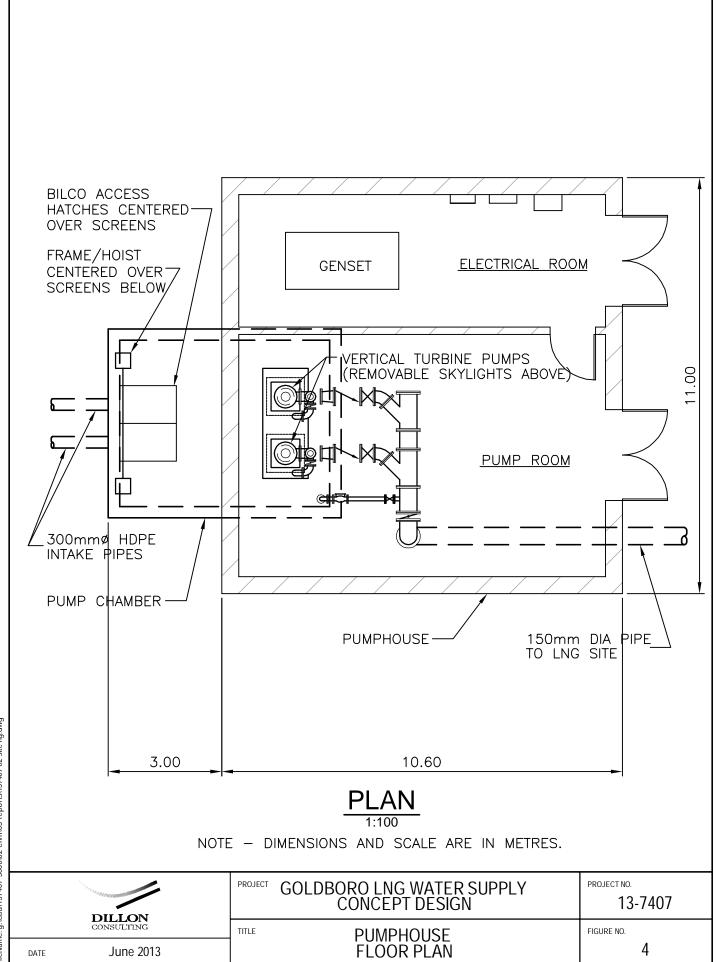
Figure 5 Site Plan



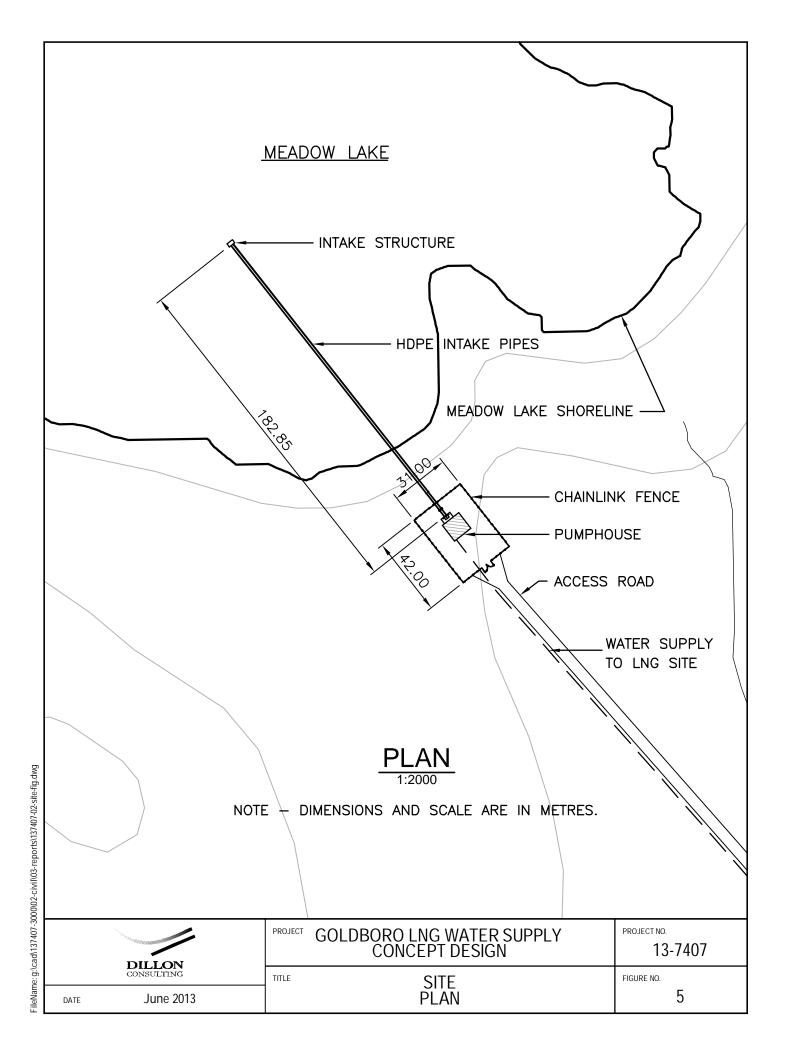
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Appendix B-3: Preliminary Chemical Inventory

Appendix B-3: Premlinary Chemical Inventory

								Human toxicity		
EC categor	Substance	CAS	UN number	Concentration (%)	Threshold amount (tonnes)		Acute	Chronic	aquatic toxicity	amount stored at Site
	acetylene	74-86-2	1001	0.01	4.5	Physical state: gas, colour: colourless, odour: ether/garlic, critical temperature: 35.3 deg C, specific garvity 0.9 (air), stability: unstable and reactive ¹	Acts as an asphyxiant, expansion may cause burns ¹	May cause damage to the following organs: lungs, upper respiratory tract, central nervous system (CNS) ¹	NA	less than threshold
	liquefied natural gas	8006-14-2	1972	0.01	4.5	Physical state: liquid, colour: colourless, odour: none, critical temperature: 537 deg C, specific garvity 0.4 (water), stability: stable ²	Liquid or cold gas contact with skin or eyes could cause freezing or severe cryogenic burns. After vapourization, contact with burning gas may cause burns. Inhalation produces weak depressant effects on the CNS at high gaseous concentrations approaching the lower flammability limit. ² Occupational limit 1000 ppm	NA	NA	less than threshold
	propane	74-98-6	1978	0.01	4.5	Physical state: gas, colour: colourless, odour: gasoline, critical temperature: 96.6 deg C, specific garvity 1.6 (air), stability: stable ³	Inhalation of this product may cause dizziness, an irregular heartbeat, narcosis, nausea or asphyxiation. Occupational limit 1000 ppm ³	NA	NA	less than threshold
	gasoline (motor fuel)	86290-81-5	1203	0.01	150	Physical state: liquid, colour: colourless, odour: gasoline, critical temperature: 257 deg C, specific garvity 3-4 (air), stability: stable 4	Inhalation of this product may cause respiratory tract irritation. Inhalation of this product may cause respirator tract irritation and Central Nervous System (CNS) Depression, symptoms of which may include; weakness dizziness, slurred speech, drowsiness, unconsciousness and in cases of severe overexposure; coma and death. Occupational limit 300 ppm. ⁴	This product contains an ingredient or ingredients, which have been shown to cause chronic toxic effects. Repeated or prolonged exposure to the	NA	less than threshold
likely to explode	benzene	71-43-2	1114	0.01	10	Physical state: liquid, colour: colourless, odour: sweetish aromatic, critical temperature: 498 deg C, specific garvity 2.8 (air), stability: stable ⁵	Acute exposure to high levels of benzene affects the central nervous system, causing dizziness, nausea, vomiting, headache, and drowsiness. HC limit -1300 ug/m ³ ⁶	In mammals, the major chronic effect is manifested on cells from the blood system. Chronic low-level exposure in humans results in blood disorders, such as pancytopenia, a reduction in the three types of formed elements in blood (erythrocytes, leucocytes, and platelets). Benzene is a known carcinogen, specifically causing acute myelogenous leukemia. In mammalian tests (e.g., rats and mice), the carcinogenicity of benzene shows a wide range of tissue and species sensitivity, with the most common effect being tumorigenicity. HClimit- 3 ug/m³ 7	NA	less than threshold
	methane	74-82-8	1971 & 1972	0.01	4.5	Physical state: gas, colour: colourless, odour: sweet, critical temperature: -82.4 deg C, specific garvity 0.422 (air), stability: stable and reactive with oxidizing agents	Inhalation of this product may cause dizziness, an irregular heartbeat, narcosis, nausea or asphyxiation. Contact with rapidly expanding gas may cause burns or frostbite. Occupational limit of 1000 ppm ⁸	NA	NA	less than threshold
	ethane	74-84-0	1035 & 1961	0.01	4.5	Physical state: gas, colour: colourless, odour: sweet, critical temperature: 32.4 deg C, specific garvity 1.1 (air), stability: stable and reactive with oxidizing agents ⁹	Inhalation of this product may cause dizziness, an irregular heartbeat, narcosis, nausea or asphyxiation. Contact with rapidly expanding gas may cause burns or frostbite. Occupational limit of 1000 ppm ⁹	NA	NA	less than threshold
	ethylene	74-85-1	1038 & 1962	0.01	4.5	Physical state: gas, colour: colourless, odour: waxy, critical temperature: 10.0 deg C, specific garvity 0.61 (air), stability: stable and reactive with oxidizing agents ¹⁰	Inhalation of this product may cause dizziness, an irregular heartbeat, narcosis, nausea or asphyxiation. Contact with rapidly expanding gas may cause burns or frostbite. Occupational limit of 1000 ppm 10	Not carcinogenic, possibly mutangenic, causes long term damage to the heart and kidney. ¹⁰	NA	less than threshold

Appendix B-3: Premlinary Chemical Inventory

EC catego	ry Substance	CAS	LIN number	Concentration (%)	Threshold amount (tonnes)	Properties	Acute	Human toxicity IChronic	aquatic toxicity	amount stored at Site
	mercury	7439-97-6	2809	N/A	1	Mercury is a dense silver-white metal that is liquid at room temperature and is characterized by low electrical resistivity, high surface tension, and high thermal conductivity. The two properties	Mercury, in its elemental form, is a liquid metal that can form a vapour at room temperature. Inhaling mercury vapours or ingesting mercury can cause	The health effects of low-level exposure include neurological damage, reproductive system damage, behavioural problems and learning disabilities. TDI = 0.003 mg/kg/day; TC = 0.00006 mg/m ³ 11	aquatic toxicity	less than threshold
	hydrogen sulphide	rogen sulphide 7783-06 1053 0.1 4.5 liquid: light h than a mixtu		Pure hydrogen sulphide is slightly heavier than air and may collect in low spots under certain conditions. H_2S is soluble in water and hydrocarbons. Any H_2S gas that comes out of solution will collect in the head spaces of tanks, pipes and vessels containing sour liquids, liquid sulfur and solid sulfur. When H_2S is mixed with some light hydrocarbons such as methane, the mixture can be lighter than air. When mixed with heavier hydrocarbons, such as NGL, the mixture is much heavier than air. Pure hydrogen sulphide is flammable in air at 40,000 ppm (4 per cent) or higher. 12	million parts of air (100 ppm) is Immediately Dangerous to Life and Health (IDLH) for exposures of 30 minutes or more. Exposure to more than 10	NA	NA	less than threshold		
inhalation	methyl mercaptan 74-93-1		1064	0.1	4.5	Physical state: gas, colour: colourless, odour: rotten cabbage, critical temperature: 196.9 deg C, specific garvity 1.6 (air), stability: stable and reactive with oxidizing agents 13	May be fatal if inhaled, occupational limit of 0.5 ppm	NA	NA	less than threshold
	sulphur dioxide	2025884	1079	0.1	2.27	Sulphur dioxide (SO ₂) is a colorless gas that smells like burnt matches. It can be chemically transformed into acidic pollutants such as sulpfuric acid and sulpfates (sulphates are a major component of fine particles). SO ₂ is generally a byproduct of industrial processes and burning of fossil fuels. ¹⁴	Inhaling sulfur dioxide is associated with increased respiratory symptoms and disease, difficulty in breathing, and premature death. 8-hour limit of 300 ug/m ^{3 15}	Inhaling sulfur dioxide is associated with increased respiratory symptoms and disease, difficulty in breathing, and premature death. Annual limit of 60 ug/m ³ 15	NA	less than threshold
	nitric oxide (nitrogen monoxide)	10102-43-9	1660	0.1	4.5	Physical state: gas, colour: colourless, odour: irritating odour, critical temperature: -93.15 deg C, specific garvity 1.04 (air), stability: stable and reactive with oxidizing agents ¹⁶	Severely irritating to eyes. Risk of serious damage to eyes. Contact with rapidly expanding gas may cause burns or frostbite. Irritating to skin. Contact with rapidly expanding gas may cause burns or frostbite. Very toxic by inhalation. Occupational limit of 25 ppm. ¹⁶	May cause damage to the following organs: blood, lungs, mucous	NA	less than threshold
	nitrogen dioxide	10102-44-0	1067	0.1	1.13	Nitrogen dioxide (NO ₂) and nitric oxide (NO), collectively referred to as nitrogen oxides (NO _x), are emitted to the atmosphere from high-temperature combustion processes (car engines, power plants, industrial processes). Although primarily emitted as NO, NO ₂ is rapidly formed when NO reacts with ozone (O ₃) and volatile organic compounds (VOC). The major sources of NO ₂ in Canada are on-road and off-road vehicles, the oil and gas industry and the use of fuel for electricity generation and heating. ¹⁷	NO2 has adverse health effects. It can irritate the lungs, decrease lung function and increase susceptibility to allergens for people with asthma. 1-hour limit 400 ug/m ³ 15	NO2 has adverse health effects. It can irritate the lungs, decrease lung function and increase susceptibility to allergens for people with asthma. Annual limit 100ug/m ^{3 15}	NA	less than threshold

Appendix B-3: Premlinary Chemical Inventory

		Human toxicity								
EC categor	y Substance	CAS	UN number	Concentration (%)	Threshold amount (tonnes)	Properties	Acute	Chronic	aquatic toxicity	amount stored at Site
non- scheduled substances	methyl diethanolamine (MDEA)	105-59-9	NA	NA	NA	Physical state: liquid, colour: colourless, odour: irritating odour, critical temperature: 138 deg C, specific garvity 4.01 (air), stability: stable and reactive with halogenated solvents ¹⁸	May cause severe irritation with corneal injury which may result in permanent impairment of vision, even blindness. Chemical burns may occur. Prolonged contact may cause moderate skin irritation with local redness. Repeated contact may cause skin burns. Symptoms may include pain, severe local redness, swelling, and tissue damage. May cause more severe response if skin is abraded (scratched or cut). May cause more severe response on covered skin (under clothing, gloves). ¹⁸		Bioconcentration potential is low (BCF less than 100 or log Pow less than 3). Potential formobility in soil is very high (Koc between 0 and 50).	
	petroleum oil lubricants		NA	NA	NΙΔ	Physical state: viscous liquid, colour: black, odour: petroleum odour, specific garvity 0.8 (water), stability: stable ¹⁹	No known significant effects or critical hazards ¹⁹	No known significant effects or critical hazards	NA	600 m ³
	Glycol	107-21-1	NA	NA		Physical state: liquid, colour: black, odour: odourless, specific garvity 1.118 (water), vapour density 3.65 (air), stability: stable ²⁰	No known significant effects or critical hazards ²⁰	In humans, effects have been reported on the following organs: Kidney. Gastrointestinal tract. In humans, symptoms may include: Headache. Nausea and/or vomiting. Abdominal discomfort ²⁰	NA	to be determined
	Diesel oil	64742-81-0	NA	NA	NA	Physical state: bright oily liquid, colour: clear to yellow, odour: mild petroleum, specific garvity 0.8 (water), vapour density 4.5 (air), stability: stable, reacts with oxidizing agents and acids ²¹	Inhalation of this product may cause respiratory tract irritation and Central Nervous System (CNS) Depression, symptoms of which may include; weakness, dizziness, slurred speech, drowsiness, unconsciousness and in cases of severe overexposure; coma and death. ²¹	Diesel engine exhaust particulate is probably carcinogenic to humans (IARC Group 2A) ²¹	NA	200 m ³
	tri-sodim phosphate	10101-89-0	NA	NA	NΙΔ	Physical state: soilid, colour: white powder, odour: NA, stability: stable, reacts with acids and magnesium ²¹	Inhalation of product may aggravate existing chronic respiratory problems such as asthma, emphysema or bronchitis. Skin contact may aggravate existing skin disease.	Inhalation of product may aggravate existing chronic respiratory problems such as asthma, emphysema or bronchitis. Skin contact may aggravate existing skin disease.	NA	10 tonnes

NA - Not available in literature search

¹ Air Liquide MSDS sheet 2011

² Fortis BC MSDS sheet 2011

Air Liquide MSDS sheet 2011
 Petro-Canada MSDS sheet 2012

⁵ Petro-Canada MSDS sheet 2012

⁶ Health Canada. 2009. Guidelines for Drinking Water Quality Guideline Technical Document

⁷ Canadian soil quality guidelines for the protection of human and environmental health. 2004. Benzene.

Canadian soil quality guidelines for the protection
 AirLiquide MSDS sheet 2011

⁹ Air Liquide MSDS 2011

¹⁰ Air Liquide MSDS 2011

¹¹ Canadian soil quality guidelines for the protection of human and environmental health. 1999. Mercury.

¹² CAPP Occupational Health and Safety of H₂S. 2003.

¹³ Air Liquide MSDS sheet 2011

¹⁴ http://www.ec.gc.ca/toxiques-toxics/default.asp?lang=En&n=22B1EA72-1

¹⁵ http://novascotia.ca/just/regulations/regs/envairqt.htm

¹⁶ Airgas MSDS 2013

¹⁷ http://www.ec.gc.ca/indicateurs-indicators/default.asp?lang=en&n=C8BFC3F2-1

¹⁸ Dow Chemical MSDS sheet 2007

¹⁹ Petro-Canada MSDS sheet 2010

²⁰ MEGlobal MSDS sheet 2013

Petro-Canada MSDS 2013cleartech MSDS 2011